

IAP12 Rec'd PCT/PTO 19 JUN 2006

**A method for separating, extracting and purifying Poly- $\beta$ -hydroxyalkanoates (PHAs) directly from bacterial fermentation broth**

**Substitute Specification - Marked-up Copy  
(As Previously Amended under PCT Article 34)**

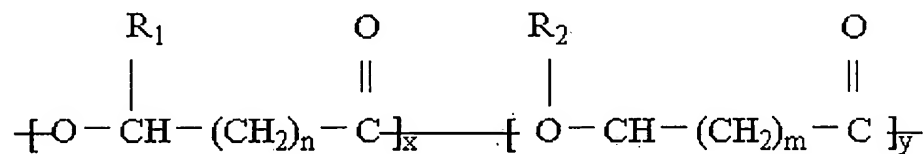
**Technical field**

This invention relates to post treatment of biological engineering, particularly to extraction and separation of bacterial fermentation products, ~~product~~, or more particularly to extraction and separation of polyhydroxyalkanoates in cells.

**Background of invention**

Poly- $\beta$ -hydroxyalkanoates (PHAs) are biological polyesters accumulated in cells by special microorganisms under special growth conditions.

General formula: The general formula of poly- $\beta$ -hydroxyalkanoates is:



In which, n and m are integers from 1 to 4, and typically 1-4 integer, ~~usually it is 1~~, that is for example 3-hydroxyalkanoates (3-HAS); R<sub>1</sub> and R<sub>2</sub> are straight chain or branched chain C<sub>1-12</sub> alkyl groups which are substituted or non-substituted; X and Y are not 0 simultaneously, and determine the content of the component in copolymer. The average molecular weight of PHAs is generally 1-4 million Da.

The physical property of PHAs is similar to that of polypropylene. As its biodegradability, biocompatibility, ~~piezoelectricity~~ piezoelectric properties and optical activity are characteristics not possessed by common petrochemical resins, it has wide application ~~prospect~~ prospects in industry, agriculture, medicine, sanitation, food, electronics, etc.

Large scale industrialized production of PHAs has not been realized internationally. The principal reason is because the cost is much higher than that of

petrochemical resin. The cost of PHAs includes mainly material cost and separation purification cost. The material cost depends on production efficiency of bacteria species and fermentation technology, whilst the separation purification cost depends largely on technology. The current extraction technology includes separation of cells from fermentation liquid with high speed centrifuge and purification of PHAs in separated wet bacterial body with organic solvent extraction, chemical reagent or surfactant + enzyme. These methods suffer from ~~have the defect of~~ high cost or serious pollution, and are difficult to be industrialized. ~~One~~ A one step extraction separation method for extracting polyhydroxyalkanoates directly from fermentation liquid containing cells is disclosed in Chinese patent application CN1328160A, but it must use large quantities ~~quantity~~ of sodium hypochlorite and has the defect of poor operation environment, serious pollution, cost increased by waste water treatment, and product quality affected by shear degradation of PHAs.

~~The~~ Chinese patent No. CN119067A discloses a method for separating and purifying polyhydroxyalkanoates in bacterial cells from bacterial thallus. The method includes two steps in extraction, which requires separation of bacterial thallus from fermentation liquid. As the bacterial thallus is very small (several microns) and the fermentation liquid has certain viscosity, it is necessary to use high speed centrifuge of separation factor higher than 6000. This greatly increases ~~greatly~~ the investment and is a bottleneck in large scale industrialized production. The method also requires expensive protease to treat the separated product which increases cost.

The extraction method disclosed in Chinese patent CN1171410A requires high speed centrifuge and lyophilization of product separated with centrifuge. These are very difficult in large scale industrialized production. Taking a factory producing 100,000 tons of PHAS/year as example, nearly 3000-4000 tons of fermentation liquid/day will be centrifuged and 600-800 tons will be lyophilized. The industrial cost will be very high. The final washing with organic solvent pollutes the environment and increases product cost.

U.S. Patent No 4,910,145 ~~US patent US4910145A~~ discloses a method for separating and extracting PHAS with enzyme and surfactant. Since the cell wall and membrane are very complicated in composition and cannot be completely decomposed

with one enzyme, it is necessary to use several enzymes or a compound enzyme. As the optimal action conditions as pH, temperature, etc. of different enzymes are different, the technology is very complicated. The method requires heating fermentation liquid to  $>80^{\circ}\text{C}$  which consumes enormous energy. The price of enzyme preparation is high, so the separation cost is high. Besides, the purity of product is not high.

The purpose of this invention is to provide an extraction method for PHAs, which can ~~reduce~~ effectively reduce separation and purification cost, reduces pollution, and is suitable for industrialized production.

## **Invention**

This invention provides a method for directly separating and purifying polyhydroxyalkanoates in cells from bacterial fermentation liquid, which comprises:  
comprisings:

- (1) pretreating fermentation liquid with a physical method for breaking cell wall;
- (2) adjusting the pH ~~value~~ of the pretreated fermentation liquid so that it is alkaline;
- (3) adding anionic surfactant and agitating;
- (4) separating and extracting coagulated precipitate in reaction liquid;
- (5) washing and drying.

The physical method includes mechanical breaking and ultrasonic breaking. The sequence of adjusting pH and adding surfactant is interchangeable. The mechanical breaking includes ball milling or high pressure homogenization.

In step 3, aside from adding anionic surfactant, coagulating agent can be added.

The mechanical breaking method used to break cell wall can be ball milling.

The pH of the pretreated fermentation liquid is adjusted to 8-13. The alkaline substance used in adjusting pH can be solid or an aqueous solution of NaOH,  $\text{Na}_2\text{CO}_3$ ,  $\text{NaHCO}_3$  or ammonia water.

The anionic surfactant can be olefinesulfonate (AOS), a fatty alcohol sulfate, a fatty alcohol polyoxyethylene-ether sulfate (AES), a fatty alcohol-polyoxyethylene ether (AEO), alkylphenol-polyoxyethylene ether, ~~etc., its~~ etc. The quantity of anionic surfactant is 0.5-20% (W/V) of fermentation liquid.

The coagulating agent is sodium polyacrylate, modified starch, polyamine, etc., its etc. The quantity of the coagulating agent is 0.5-20% (W/V) of the fermentation liquid.

After adding the anionic surfactant and the coagulating agent, the reaction temperature under agitation is 10-70°C and the reaction time 5-60min.

Centrifuge, filter-press, vacuum suction filtration, etc. can be used for separating and extracting coagulated precipitate from the reaction liquid.

The invention is applicable to separation and purification of fermentation liquid of bacteria and its aberrance and gene engineering bacteria containing polyhydroxyalkanoates. Applicable bacteria include Alcaligenes, Pseudomonas, Azotobacter, Rhodospirillum, Methylophilus, Bacillus, etc.

The invention has no high requirement for dry weight of cells and content of PHAs in fermentation liquid. The invention has the advantage of simple technology, low cost, high yield and greatly reduced pollution, so large scale industrialized production can be realized.

#### Detailed description of the invention

The following ~~Following~~ examples are used to further describe the invention. These examples should not be construed as limiting ~~constitute any limitation to~~ the scope of claimed invention. ~~claims~~. Any modifications or changes made by the skilled man in the art benefited ~~benefit~~ from the disclosure of this application should be included within the scope of claims stated in this application.

#### Example 1

~~Take~~ 1000ml of fermentation liquid of Alcaligenes entrophus mutant 65-7, in which the dry weight of cells ~~is 142g/l, was 142g/l~~ and the content of PHBV ~~is 78.5%; pretreat was 78.5%~~ was pretreated with a ball mill (530r/min, 0.1mm steel ball) for ~~40min; adjust 40 min.~~ The pH value of the resulting solution was adjusted to 12 with 30% NaOH solution. ~~solution; add 13g of sodium laurylsulfate; adjust reaction~~ laurylsulfate was added to the solution and the temperature was adjusted to 32°C; react under 32°C while the solution was subjected to agitation for 5min; filter 5 min. ~~The~~

solution was filtered with suction and filter paper; wash paper. The coagulated precipitate was washed with water till until the washing becomes neutral; dry at 70°C to constant weight. Purity liquid became neutral, then dried at 70°C. The purity of the product is was 98.2%. The 98.2%, the average molecular weight is was  $5.2 \times 10^5$  Da. The  $5.2 \times 10^5$  Da, the yield is was 85.2%. the The COD and BOD of waste water from suction filtration after treatment with anaerobic and aerobic bacteria is 800 and 30mg/l respectively, in conformity with state discharge standards. standard.

### **Example 2**

Take 100ml of fermentation liquid of Alcaligenes entrophus, in which the dry weight of cells is 147g/l, was 147g/l and the content of PHBV is 75.2%; break-cell-wall was 75.2% was subject to a process that broke the cell walls with ultrasonic (1500W) for 20min; adjust 20 min. The pH value of the resulting solution was adjusted to 8 with 30% NaOH solution; add solution. 0.5g of sodium laurylsulfate and 5g of sodium polyacrylate were added to the solution and the polyacrylate; adjust reaction temperature of the solution was adjusted to 70°C; react under 70°C while the solution was subjected to agitation for 30min; filter 30 min. The solution was filtered with suction and filter paper; wash paper. The coagulated precipitate was washed with water till until the washing becomes neutral; dry in oven at 70°C to constant weight. Purity liquid became neutral, then dried at 70°C. The purity of the product was 93.2%. The is 93.2%, the average molecular weight is  $4.1 \times 10^5$  Da, the yield is 80.3%. was  $4.1 \times 10^5$  Da. The yield was 80.3%.

### **Example 3**

Take 50ml of fermentation liquid of Alcaligenes entrophus, in which the dry weight of cells is 102g/l, was 102 g/l and the content of PHB is 60%; pretreat was 60% was pretreated with a ball mill (560r/min, 0.1mm steel ball) for 30min; adjust 30 min. The pH value of the solution was adjusted to 13 with  $\text{NH}_3 \cdot \text{H}_2\text{O}$  solution; add solution. 10g of sodium laurylsulfate and 10g of modified starch; adjust reaction starch were added to the solution and the temperature of the solution was adjusted to 10°C; react under 10°C while the solution was subjected to agitation for 10min; separate with centrifuge 10 min. The solution was centrifuged (separation factor 600); wash— 600) and the coagulated

~~precipitate was washed with water till until the washing becomes neutral; dry in oven at 70°C to constant weight. Purity liquid became neutral, then dried at 70°C. The purity of the product is 98.2%, the was 98.2%. The average molecular weight is  $4.4 \times 10^5$  Da, the was  $4.4 \times 10^5$  Da. The yield is was 87%.~~

#### Example 4

~~Take 500ml of fermentation liquid of Alcaligenes entrophus mutant 65-7, in which the dry weight of cells is 135g/l, was 135 g/l and the content of PHB is 75.5% and introducee was 75.5% was introduced it into a special vessel. Increase The pressure in the vessel was increased to 60MPa, release pressure 60 MPa and rapidly released after 10min, collect 10 min and the liquid was collected. This operation was repeated and repeat the operation twice. Adjust The pH value of the solution was adjusted to 10 with 30% NaOH solution; add solution. 9g of sodium lauryl polyoxyethylene ether sodium sulfate; adjust reaction sulfate was added to the solution and the temperature of the solution was adjusted to 38°C; react under 38°C while the solution was subjected to agitation for 8min; filter 8 min. The solution was filtered with suction and filter paper; wash paper. The coagulated precipitate with water till washing becomes neutral; dry at 70°C to constant weight. Purity was washed with water until the washing liquid became neutral. The purity of the product is 96.7%, the was 96.7%. The average molecular weight is  $4.2 \times 10^5$  Da, the was  $4.2 \times 10^5$ . The yield is was 81.5%.~~

#### Example 5

~~Take 100ml of fermentation liquid of Alcaligenes entrophus, in which the dry weight of cells is 154g/l, was 154 g/l and the content of PHBV is 80.5%; break cell wall was 80.5% was subject to a process that broke the cell walls with ultrasonic (2800W, continuous treatment) for 40min; adjust 40 min. The pH value of the solution was adjusted to 11 with 30% NaOH solution; add solution 10kg of sodium lauryl sulfate and 0.5kg of sodium polyacrylate; adjust reaction polyacrylate was added to the solution and the temperature to 50°C; react under 50°C while the solution was subjected to agitation for 60min; filter 60 min. The solution was filtered with a filter press; wash press. The coagulated precipitate was washed with water till until the washing became neutral and~~

~~the filtrate was dried becomes neutral; dry~~ in oven at 70° to a constant weight. ~~Purity~~ The  
purity of the product is 97%, the was 97%. The average molecular weight was  $5.3 \times 10^5$   
Da. The is  $5.3 \times 10^5$  Da, the yield is was 84%.

### Example 6

~~Take~~ 100ml of fermentation liquid of Pseudomonas, in which dry weight of cells was 86  
g/l and the is 86g/l, content of PHBV is ~~61.5%~~; ~~pretreat~~ was 61.5% was pretreated with a  
ball mill (560r/min, 0.1mm steel ball) for ~~50min~~; ~~adjust~~ 50 min. The pH value of the  
solution was adjusted to 11 with 30% NaOH ~~solution~~; ~~add~~ solution. 3g of sodium  
~~laurylsulfate~~; ~~adjust reaction~~ laurylsulfate was added to the solution and the temperature  
of the solution was adjusted to 24°C; ~~react under~~ 24°C while the solution was subjected  
to agitation for 10min; ~~filter~~ 10 min. The solution was filtered with suction and filter  
~~paper~~; ~~wash~~ paper. The coagulated precipitate was washed with water ~~til~~ until the  
washing becomes neutral; dry became neutral. The filtrate was dried at 70°C to a  
constant weight. ~~Purity~~ The purity of the product is 94.2%, the was 94.2%. The average  
molecular weight is  $3.2 \times 10^5$  Da, the was  $3.2 \times 10^5$  Da. The yield is was 71.2%.